



Form SP2

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UNIVERSITY OF SARAJEVO – FACULTY OF SCIENCE Department of Chemistry

| Course ID: HODOA21 | | rse name: DEVELOPMENT AND APPLICATION OF METAL MPLEX - SELECTED TOPICS | | | |
|---------------------------------------------------------------------------------------------------------------------------------------|------------------------------------------------|---------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|--|--|--|
| Cycle: THIRD | Year: FIRST | Semester: II ECTS credits: 10 | | | |
| Course status: ELEC | CTIVE | Total course hours: 60 Lectures: 30 Laboratory: 30 | | | |
| Teaching participa | nts: Teachers Chemistry | with expertise in the field of Inorganic | | | |
| Prerequisite for enrollment: | - | | | | |
| Course aims: | 1 0 | Developing the ability to design new complex compounds, depending on the possible application | | | |
| 1. Developm relationship 2. Tumor in nonplatinum 3. The mech 4. Ruthenium compounds 5. The interabiomolecule 6. Research 7. Metal com | | ent of metal complexes as potential drugs: the between synthesis-structure-properties. nibiting metal complexes. Platinum and complexes. anism of action of tumor inhibitory complex. In complexes, and complexes are complexes. Development and design of new actions of metal ions and metal complexes with | | | |
| Learning outcomes | and critical — meta — ratio targe — design | After the course the student will be able to present, discuss and critically review: - metal containing drugs and structure-activity relationship - rationalize design of metal complexes toward specific targets - design and synthesis of biomimetic models and rationalize design of catalysts | | | |
| Teaching methodo | | Auditory lectures, discussion, critical review | | | |

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| | Grading criteria | | | | |
|------------------------------------|---------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|--------------------|----------------|--|--|
| | Crit | eria Maximal score | Required score | | |
| | 1. Tests | 2×15 | 16,5 | | |
| | 2. Seminars | 1×30 | 16,5 | | |
| | 3. Final exam | 40 | 22 | | |
| | Total | 100 | 55 | | |
| Assessment methods | Scores and grading | | | | |
| and grading system ¹ : | Score | Grade | Grade | | |
| anu gi aunig system ² . | | (B&H) | (ECTS) | | |
| | < 55 | 5 | F, FX | | |
| | 55-64 | | <u>E</u> | | |
| | 65-74 | | <u>D</u> | | |
| | 75-84 | | <u>C</u> | | |
| | 85-94 95-100 | | <u>В</u> А | | |
| | | | | | |
| Literature ² : | Mandatory literature: Original scientific papers Metal Complexes in Cancer Chemotherapy, edited by B. Keppler, VCH, 1993 C. Jones, J. Thornback, Medicinal Applications in Coordination Chemistry RSC, 2007 Metal Complexes: DNA interactions, edited by Nick Hadjiliadis, Einar Sletten, Wiley, 2009 | | | | |

 1 The grading structure for each subject is determined by the Council of the organizational unit before the beginning of the academic year in which the subject is taught as per Article 64, paragraph 6 of the Law on Higher Education of Sarajevo Canton

² The Senate of the higher education institution, as an institution, or the Council of the organizational unit of the higher education institution, as a public institution, determines by a special decision, which is published on its website before the beginning of the academic year obligatory, mandatory and recommended textbooks and manuals, as well as other recommended literature based on which exams are prepared and taken as per Article 56, paragraph 3 of the Law on Higher Education of the Sarajevo Canton