



Form SP2

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UNIVERSITY OF SARAJEVO – FACULTY OF SCIENCE Department of Chemistry

Course ID: HODNM22	Course name: EVALUATION IN CHEMISTRY EDUCATION				
Cycle: THIRD	Year: FIRST	Semester: II	ECTS credits: 10		
Course status: ELEC	•	Total course hours: 60 Lectures: 30 Laboratory: 30			
Teaching participa		Teachers and associates with expertise in the field to which the subject belongs			
Prerequisite for enrollment:	-				
Course aims:	teaching. Understandi	Enabling students for independent evaluation in chemistry teaching. Understanding the impact of internal and external evaluation on the process of ensuring quality of chemistry education.			
Thematic course ur	2. Basic pri chemistr 3. Evaluation 4. Evaluation Bloom's 5. Assessm 6. Evaluation 7. Evaluation 8. Standard 9. Evaluation 10. Character program	 Evaluation of student achievements in chemistry teaching Basic principles of evaluation of achievements in chemistry Evaluation of preconceptions Evaluation of tests and student achievements according to Bloom's taxonomy Assessment in chemistry teaching Evaluation of textbooks and tests of knowledge Evaluation of the chemistry curriculum Standardized tests Evaluation of chemistry teachers Characteristics of the most important international programs for the evaluation of student achievements in science education 			
Learning outcomes	Skills: Competence Evaluatopic Analy	Knowledge:			
Teaching methodol	Discussion				

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	Grading criteria				
	l <u> </u>	Criteria	Maximal score	Required score	
	1.	Class attendance	-		
	2.	Class activities	-		
	3.	Midterm	20	11	
	4.	Seminar	40	22	
	5.	Final exam	40	22	
Assessment methods		Total	100	55	
and grading system ¹ :	Scores and grading				
and grading system.	Score		Grade	Grade	
		< 55	(B&H)	(ECTS)	
		<u> </u>	<u>5</u> 6	F, FX E	
		65-74			
	-	75-84	8	C	
		85-94	9	В	
		95-100	10	A	
	Mand		10		
	Mandatory literature:				
	1. Anderson, L. W. et.al. (2000). A Taxonomy for Learning,				
	Teaching, and Assessing: A Revision of Bloom's				
	Taxonomy of Educational Objectives. 2nd Edition. New				
	York: Pearson Allyn and Bacon.				
	2. Liu, X. (2010). Essentials of Science Classroom				
Literature ² :	Assessment. Thousand Oaks: SAGE.				
	3. McMahon, M., Simmons, P., Sommers, R., De Baets, D.,				
	& Crawley, F. (2006). Assessment in Science: Practical				
		Experiences and	Educational Resea	rch. Arlington:	
		NSTA.		S	
	⊿.		aghan Τ (200Ω) Δcc	essina National	
	4. Greaney, V., & Kellaghan, T. (2008). Assessing National				
	Achievement Levels in Education. Washington: The				
		World Bank.			

 $^{^{}m 1}$ The grading structure for each subject is determined by the Council of the organizational unit before the beginning of the academic year in which the subject is taught as per Article 64, paragraph 6 of the Law on Higher Education of Sarajevo Canton

 $^{^2}$ The Senate of the higher education institution, as an institution, or the Council of the organizational unit of the higher education institution, as a public institution, determines by a special decision, which is published on its website before the beginning of the academic year obligatory, mandatory and recommended textbooks and manuals, as well as other recommended literature based on which exams are prepared and taken as per Article 56, paragraph 3 of the Law on Higher Education of the Sarajevo Canton